## Diels-Alder Addition of Cyclopentadiene to Allenic Esters: Catalysis by Lanthanide Complexes

## R. P. Gandhi,\* M. P. S. Ishar, and A. Wali

Department of Chemistry, Indian Institute of Technology-Delhi, Hauz Khas, New Delhi-110016, India

Stereoselectivity and yields are markedly enhanced for Diels–Alder additions involving cyclopentadiene and allenic esters (**1a**—c), on catalysis by lanthanide complexes, Eu(fod)<sub>3</sub> and Pr(fod)<sub>3</sub>.

Table 1

In general, the use of allenes as synthons in cycloadditions is restricted on account of their low dienophilicity and tendency to undergo excessive polymerisation under forcing conditions.<sup>1</sup> In this communication, we report cycloaddition reactions involving cyclopentadiene and three allenic esters *i.e.*, ethyl penta-2,3-dienoate (**1a**), ethyl hexa-2,3-dienoate (**1b**), and ethyl 4-methylpenta-2,3-dienoate (**1c**), with and without catalysis<sup>2</sup> by Eu(fod)<sub>3</sub> and Pr(fod)<sub>3</sub>.† Under catalytic conditions, a marked improvement in stereoselectivity and yields of the cycloadducts is observed.

In an uncatalysed reaction, 4 mM of the allenic ester‡ was refluxed with 2.5—3.0 equiv. of cyclopentadiene in 50 ml of dry benzene. In the catalysed reaction, 1.0 mole% of Eu(fod)<sub>3</sub> or  $Pr(fod)_3$  was added to the reaction mixture (both catalysts gave identical results) and left at room temperature. The products were separated using Chromatotron-2 (silica gel PF254 Merck, 2 mm, benzene–hexane 1:1, 5 ml/min.) or by column chromatography (silica gel, 60—120 mesh, petroleum ether–benzene 9:1), and were characterised by mass, <sup>1</sup>H n.m.r. (with extensive homodecoupling experiments), and <sup>13</sup>C n.m.r. (COM decoupled and SFORD)§ spectroscopy, and comparison of spectral data with that of known bicyclo[2.2.1]heptene systems.<sup>3</sup> The results are presented in Table 1.

The distinction between 2-endo and 2-exo configurations of the adducts is based on the value of vicinal coupling constant,  ${}^{3}J_{1,2}$  and the C<sub>2</sub>-H chemical shift.<sup>3a,b</sup> The assignments are further supported by the observation of long range coupling,  ${}^{4}J_{2,7s}$  (ca. 1.8 Hz), in the case of exo-adducts (2**a**-**c**) (due to W relationship),<sup>3a,4</sup> and the downfield shift of C<sub>3</sub>, C<sub>1</sub>, and in particular, C<sub>6</sub>, and the upfield shift of C<sub>4</sub>, on going from endo (3**a**,**b**,**c**) to the corresponding exo-adducts (2**a**,**b**,**c**).<sup>3a,5</sup> The assigned stereochemistry at the exo-cyclic double bond in

 $\dagger$  fod = 1,1,1,2,2,3,3-heptafluoro-7,7-dimethyloctane-4,6-dionate.

<sup>‡</sup> The esters (**1a**—c) were prepared according to literature method; R. P. Gandhi, M. P. S. Ishar, and A. Wali, *Tetrahedron Lett.*, 1987, 6679.

§ Selected spectroscopic data: (n.m.r. spectra were recorded at 99.55 MHz for <sup>1</sup>H and 24.99 MHz for <sup>13</sup>C nuclei in CDCl<sub>3</sub>; J values are in Hz): (2a)  $v_{max}$  (CCl<sub>4</sub>): 1725 (CO<sub>2</sub>-); <sup>1</sup>H n.m.r.  $\delta$  5.54 (sp q, 1H, C<sub>8</sub>-H,  $J_{8,C(8)-Me}$  6.80,  $J_{2,8}$  2.46), 2.84 (br s, 1H, C<sub>2</sub>-H), 2.01 (br d, 1H,  $C_{(7)}-H_{anti}$ ), 1.62–1.48 (m, 1H, C<sub>7</sub>-H<sub>syn</sub>,  $J_{7a, 78}$  8.40,  $J_{2, 75}$  1.85), 1.54 (sp d, 3H, C<sub>8</sub>-Me,  $J_{2, C(8)-Me}$  0.97); <sup>13</sup>C n.m.r.  $\delta$  173.6 (CO<sub>2</sub>-), 140.5 (C<sub>3</sub>), 137.6 (C<sub>6</sub>), 135.2, 116.5, 60.4, 50.1 (C<sub>4</sub>), 47.7, 47.1 (C<sub>1</sub>, C<sub>2</sub>), 46.8 (C<sub>7</sub>), 15.0, 13.8; mass (m/z): 192(30) (M<sup>+</sup>), 119(100). (3a)  $v_{max}$ (CCl<sub>4</sub>): 1730 (CO<sub>2</sub>-); <sup>1</sup>H n.m.r.  $\delta$  5.25 (sp q, 1H, C<sub>8</sub>-H,  $J_{8, C(8)-Me}$ 6.84,  $J_{2,8}$  1.85), 3.39 (m, 1H, C<sub>2</sub>-H,  $J_{1,2}$  3.75), 1.64 (br d, 1H, C<sub>7</sub>-H<sub>anti</sub>,  $J_{7a,78}$  8.59), 1.53 (sp d, 3H, C<sub>8</sub>-Me,  $J_{2,C(8)-Me}$  0.97), 1.45 (br d, 1H,  $C_{7}$ -H<sub>syn</sub>); <sup>13</sup>C n.m.r.  $\delta$  171.9 (CO<sub>2</sub>-), 139.4 (C<sub>3</sub>), 135.4, 132.7 (C<sub>6</sub>), 115.9, 59.7, 51.6 (C<sub>4</sub>), 49.7 (C<sub>7</sub>), 47.1 (C<sub>2</sub>), 46.4 (C<sub>1</sub>), 15.2, 13.8; mass (m/z): 192(5) (M<sup>+</sup>), 28(100). (4a)  $v_{max}$  (CCl<sub>4</sub>): 1725 (CO<sub>2</sub>-); <sup>1</sup>H n.m.r.  $\delta$ 5.41 (sp q, 1H, C<sub>8</sub>-H,  $J_{8,C(8)-Me}$  7.10,  $J_{2,8}$  1.49), 3.30 (br d, 1H,  $C_{2}$ -H,  $J_{1,2}$  3.75), 1.66 (sp d, 3H, C<sub>8</sub>-Me,  $J_{2,C(8)-Me}$  2.04); <sup>13</sup>C n.m.r.  $\delta$ 172.5 (CO<sub>2</sub>-), 139.7, 134.4, 132.9 (C<sub>6</sub>), 116.3, 60.1, 49.3, (C<sub>7</sub>), 48.6, 48.3 (C<sub>2</sub>, C<sub>4</sub>), 46.7 (C<sub>1</sub>), 15.5, 14.0; mass (m/z): 192 (24) (M<sup>+</sup>), 28 (100). Similar data were obtained for (**2b**, c), (**3b**, c), and (**4b**). (2a,b), (3a,b) and (4a,b) is based on: i, comparison of allylic and, in particular, homoallylic couplings involving C<sub>2</sub>-H and C<sub>8</sub>-H, and C<sub>2</sub>-H and C<sub>8</sub>-alkyl groups, respectively, in pairs of the *endo*-adducts (3a, 4a and 3b, 4b); here, a high value of homoallylic coupling supports the *trans*-arrangement between C<sub>2</sub>-H and C<sub>8</sub>-alkyl groups;<sup>3a,4</sup> ii, comparison<sup>5,6</sup> of  $\delta$  C<sub>2</sub> and C<sub>4</sub> in (3a,b), (4a,b) with corresponding values in the *endo*-adduct (3d) (R = R<sup>1</sup> = H, C<sub>2</sub> at  $\delta$  49.5 and C<sub>4</sub> at  $\delta$  51.1)¶; and iii, comparison<sup>5,6</sup> of  $\delta$  C<sub>2</sub> and  $\delta$  C<sub>4</sub> in (2a,b) with corresponding values in the *exo*-adduct (2d) (R = R<sup>1</sup> = H, C<sub>2</sub> at  $\delta$  48.6 and C<sub>4</sub> at  $\delta$  50.0).¶



Table I.				
Allenic ester	Conditions (temp.)	Reaction time/h	% Yield	Product ratio (2:3:4)
( <b>1a</b> )	Uncatalysed			
```	(benzene, reflux)	14	38	25:67:8
	Catalysed (r.t.)	4	79	15:85
(1b)	Uncatalysed			
. ,	(benzene, reflux)	14	32	23:69:8
	Catalysed (r.t.)	4	84	14:86
(1c)	Uncatalysed			
	(benzene, reflux)	10		No reaction (tlc)
	Catalysed (r.t.)	6	60	10:90

¶ Data for (2d) and (3d), unpublished results from this laboratory.

In the above uncatalysed reactions, preferred formation of the Z isomers of both *exo*- and *endo*-adducts (**2a**,**b** and **3a**,**b**) takes place, the E isomers (**4a**,**b**) being produced to a lesser extent.|| Under Eu(fod)<sub>3</sub> and Pr(fod)<sub>3</sub> catalysis, however, the formation of E isomers is totally suppressed while *endo*selectivities and the overall yield of the adducts are markedly enhanced. In addition, the latter reactions require lower temperatures and shorter reaction times.

Received, 5th April 1988; Com. 8/01339J

|| The formation of three 1:1 adducts in the uncatalysed reaction involving each of the allenic esters, (1a) and (1b), was confirmed by g.c.m.s. studies done on a Jeol JMS-D300 instrument (SE-30, 3 mm  $\times$ 3 m, 120–250 °C, 10 °C/min., helium 1.4 kg/cm<sup>2</sup>); the retention times obtained (min) were (2a) 3.5; (3a) 3.0; (4a) 2.7; (2b) 3.5; (3b) 3.2; (4b) 3.0. Other recognisable peaks in the chromatogram corresponded to cyclopentadiene, dicyclopentadiene, the esters (1a,1b) and their dimers (mass).

## References

- (a) H. Pledger, Jr., J. Org. Chem., 1960, 25, 278; (b) L. Ghosez and M. J. O'Donnel, in 'Pericyclic Reactions,' eds. A. P. Marchand and R. E. Lehr, Academic Press, New York, 1977, vol. II, p. 55; (c) H. Hopf, in 'The Chemistry of The Allenes,' ed. S. R. Landor, Academic Press, New York, 1982, vol. 2. p. 563; (d) W. Oppolozer and C. Chapius, *Tetrahedron Lett.*, 1983, 4665; (e) K. Hayakawa, H. Nishiyama, and K. Kanematsu, J. Org. Chem., 1985, 50, 512.
- 2 S. Danishefski and M. Bednarski, Tetrahedron Lett., 1985, 2507.
- 3 (a) L. M. Jackman and S. Sternhell, 'Application of N.M.R. Spectroscopy,' Pergamon Press, Oxford, 1969; (b) W. L. Nelson and D. R. Allen, *J. Heterocycl. Chem.*, 1972, **9**, 561; (c) J. B. Stothers, '<sup>13</sup>C N.M.R. Spectroscopy,' Academic Press, London, 1972.
- 4 A. Gaudemer, in 'Stereochemistry,' ed. H. B. Kagan, Georg Thieme Publishers, Stuttgart, 1977, vol. 1, pp. 45-61, 111.
  5 (a) E. Lippmaa, T. Pehk, J. Pasasivirta, N. Bellikova and A. Plate,
- 5 (a) E. Lippmaa, T. Pehk, J. Pasasivirta, N. Bellikova and A. Plate, Org. Magn. Reson., 1970, 2, 581; (b) N. K. Wilson and J. B. Stothers, in 'Topics in Stereochemistry,' eds. N. L. Allinger and E. L. Eliel, John Wiley, New York, 1974, vol. 8, p. 43.
- 6 G. Van-Binst and D. Tourwe, Org. Magn. Reson., 1972, 4, 625.